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Electrochemical behaviour of Heat treated Al 356 Alloy using N2-phenyl-1, 3, 5-triazine- 2, 4-diamine compound in 3.5% NaCl solution

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Abstract

In the present research work corrosion behavior of heat treated Al 356 alloy was studied using 3.5% NaCl solution as corrosion medium in different concentration of inhibitor N2-Phenyl-1, 3, 5-Triazine- 2, 4-Diamine. Rectangular specimen of dimensions 2cm x1cm x1mm were machined from Al 356 alloy. Specimens were subjected heat treatment for 2h, 3h, 4h, 5h & 6 hours in muffle furnace at 550 °C. Specimens were subjected to potentiodynamic polarization studies and electrochemical impedance studies using electrochemical work station model CH 608E series manufactured by CH Instruments, USA. The results obtained were compared with that of non-heat treated specimen. It was found that the heat treated specimens exhibit excellent corrosion resistance when compared to no- heated specimen.

Catalytic activity of Silver nanoparticles: a critical review

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Metal nanoparticles are preferred in industrial applications due to their stability and unique ability. One of the major areas of application of nanoparticles is catalysis due to specificity, efficiency and recovery of the catalyst. Among the metal nanoparticles, nanosilver is used in many reactions as catalyst. These nanoparticles are used in biological and chemical reactions as dye degradation, oxidation reactions, enzymatic reaction, antimicrobial activity etc. It is also observed that there is a size and concentration dependent catalytic activity of these silver nanoparticles. This review is critically focussed on the conditions and parameters affecting the catalytic nature of the silver nanoparticles. Therefore silver nanoparticles can be used as an efficient catalyst in differ reactions. Key words: Silver Nanoparticles, Catalyst, chemical and biological reaction

